S	Start here.
1	Electrolytic Cell, Brine is electrolised, which separates Na from
	Cl. Because the sodium is Mt, it goes to the cathole,
	and Chlorine I extracted from the anode. The Nat is
	then extracted and mixed with water. Nat + Az 0 > NaOH++H
_	The hydrogen is extracted as gas and the product remains
_	Nacl -> Nat + 2C/2 The electrolysis of malten
	Nall is much more pure but for more costly.
	In Brine the electrolysis also effects the conter, so
_	energy is last through cleekolysis of water.
	140 7 Hz +202
	K = [562] x [505] 60,450 = 250, +350,
	0.325 0.05 0.03
	LAT SACOURTY START
	= (SP) × (SP3)
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-	$\leq (50) \times (50)$
-	50 7 42
-	V - C/2)
-	K= 110-77

Emulsification, a anxious of ting droplets of one subtant suspended in another,	Question 32	2010 H3C Ch	emsu y	Sam _l
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Lin	estone (a Coz) is a combisted to the solving
	process to produce CaO and Cos. EaO is what
	the solvay process but it impacts the environment
V	reatly.
The	release of COs contributes to the effect of climate change
wh	release of CO, contributes to the effect of climate change ich is a huge negative impact on our environment.
_	
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1	South Could Come as Feare per head Specifical Strawnships yet the 199
	en destructive de pour d'Alacteur en la Mantempres sant sur sur la Maistre physiciste de la comme de la comme d
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