

A.

(i) Iron.

(ii) It is ~~not~~ ~~doesn't~~ doesn't react to sea water because it ~~has~~ contains a charge to allow it not to react.

B.

(i) Bronze

(ii) ~~To give them~~ To give them a charge so they won't react with the sea water.

C.

It makes the much stronger and able to be used ~~for~~ ~~used~~ a larger number of purposes.

D.

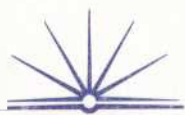
(i) corrosion - ~~oxid~~ where oxidising takes place.

~~the oxygen eats away at the metal and~~

It is when an object breaks down and falls apart.

(ii) Placing different metals in test tubes that all have water in them.

(iii) Using a different solution instead of water and make sure that they're all the same temperature and don't shake the test tube.



e. Bronze - sit it in a solution containing ~~bronze in acid and the bronze~~ acid or use vinegar and the bronze should shine up clearly.